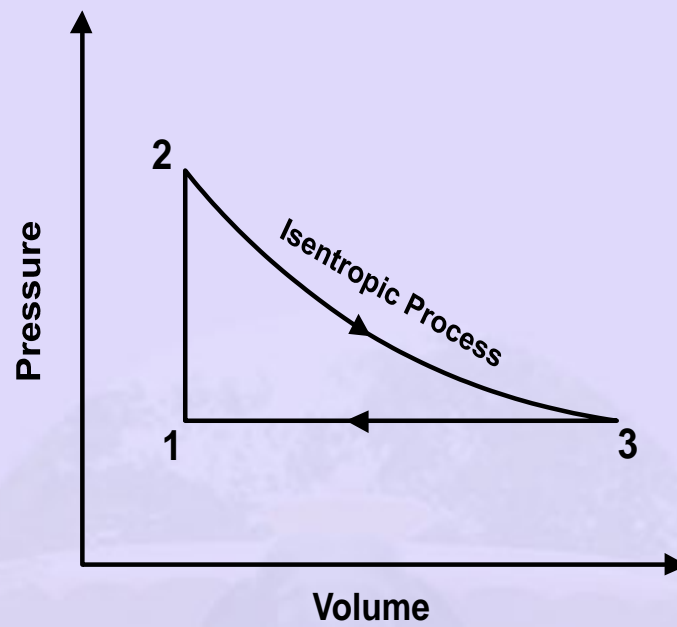
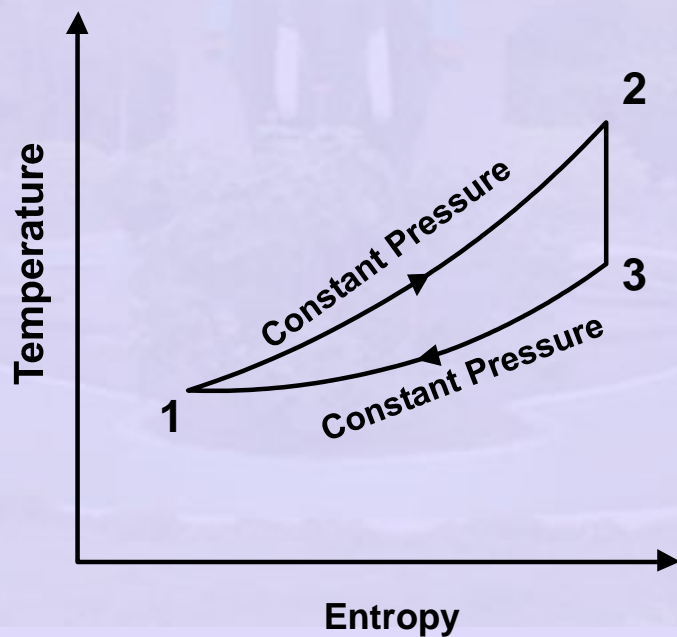


4.9 Lenoir Cycle:



(a)



(b)

Fig.4.9. Lenoir cycle on p-v and T-s diagrams

The Lenoir cycle consists of the following processes:

- *Process 1-2: Constant volume heat addition*
- *Process 2-3: Reversible adiabatic expansion*
- *Process 3-4: Constant pressure heat rejection.*
- *No compression process.*

The thermal efficiency can be derived as follows:

$$\eta_{th} = \frac{\text{Heat added} - \text{Heat rejected}}{\text{Heat added}} = \frac{C_v (T_2 - T_1) - C_p (T_3 - T_1)}{C_v (T_2 - T_1)}$$

$$\eta_{th} = 1 - \gamma \left(\frac{T_3 - T_1}{T_2 - T_1} \right)$$

$$\text{Let, } \frac{P_2}{P_1} = r_p = \text{Pressure ratio}$$

$$T_2 = r_p T_1$$

$$\frac{T_3}{T_2} = \left(\frac{P_3}{P_2} \right)^{\frac{\gamma-1}{\gamma}} = \left(\frac{P_1}{P_2} \right)^{\frac{\gamma-1}{\gamma}} = \left(\frac{1}{r_p} \right)^{\frac{\gamma-1}{\gamma}} = (r_p)^{\frac{1}{\gamma}-1}$$

$$T_3 = T_2 (r_p)^{\frac{1}{\gamma}-1} = T_1 r_p (r_p)^{\frac{1}{\gamma}-1} = T_1 (r_p)^{\frac{1}{\gamma}}$$

$$\eta_{th} = 1 - \frac{\gamma \left(\frac{1}{r_p^{\frac{1}{\gamma}}} - 1 \right)}{(r_p - 1)}$$