

## Unit 12 - Week 10

## Course outline

How does an NPTEL online course work?

Week 0

Week 1

Week 2

Week 3

Week 4

Week 5

Week 6

Week 7

Week 8

Week 9

Week 10

● Phase Diagrams (Congruent melting alloys, Peritectic Reaction, Monotectic Reaction)

○ Phase Diagrams (Allotropy, Eutectoid and Peritectoid Reactions)

○ Phase Diagrams (Iron-Iron Carbide Phase Diagram)

○ Quiz : Assignment 10

○ Week 10 Feedback Form : Basics of Materials Engineering

○ Assignment 10 solution

Week 11

Week 12

Video Download

Live Session

Text Transcripts

## Assignment 10

The due date for submitting this assignment has passed.  
As per our records you have not submitted this assignment.

Due on 2020-11-25, 23:59 IST.

1) The invariant reaction of two solids transforming to a single solid at constant temperature upon heating is known as a 1 point

- eutectic reaction  
 eutectoid reaction  
 peritectic reaction  
 peritectoid reaction

No, the answer is incorrect.  
Score: 0

Accepted Answers:  
eutectoid reaction

2) Monotectic reaction refers to 1 point

- a liquid phase transforming to another liquid and solid phase upon cooling  
 a liquid phase transforming to another liquid and solid phase upon heating  
 a liquid phase and a solid phase transforming into another solid phase upon cooling  
 two liquids transforming to another liquid phase and solid phase upon cooling

No, the answer is incorrect.  
Score: 0

Accepted Answers:  
a liquid phase transforming to another liquid and solid phase upon cooling

3) Identify the CORRECT statement(s) 1 point

- Solubility of carbon in plain carbon steel is higher in BCC ferrite since it has more free space compared to FCC austenite.  
 Solubility of carbon in plain carbon steel is higher in FCC austenite since it has single largest free volume compared to BCC ferrite.  
 The solubility of carbon in steel has nothing to do with its crystal structure.  
  $\delta$ -iron is magnetic and has BCC crystal structure.

No, the answer is incorrect.  
Score: 0

Accepted Answers:  
Solubility of carbon in plain carbon steel is higher in FCC austenite since it has single largest free volume compared to BCC ferrite.

Figure 1 shows the phase diagram of Fe-Fe<sub>3</sub>C system. Based on the phase diagram, answer the following questions (Q4 – Q11).

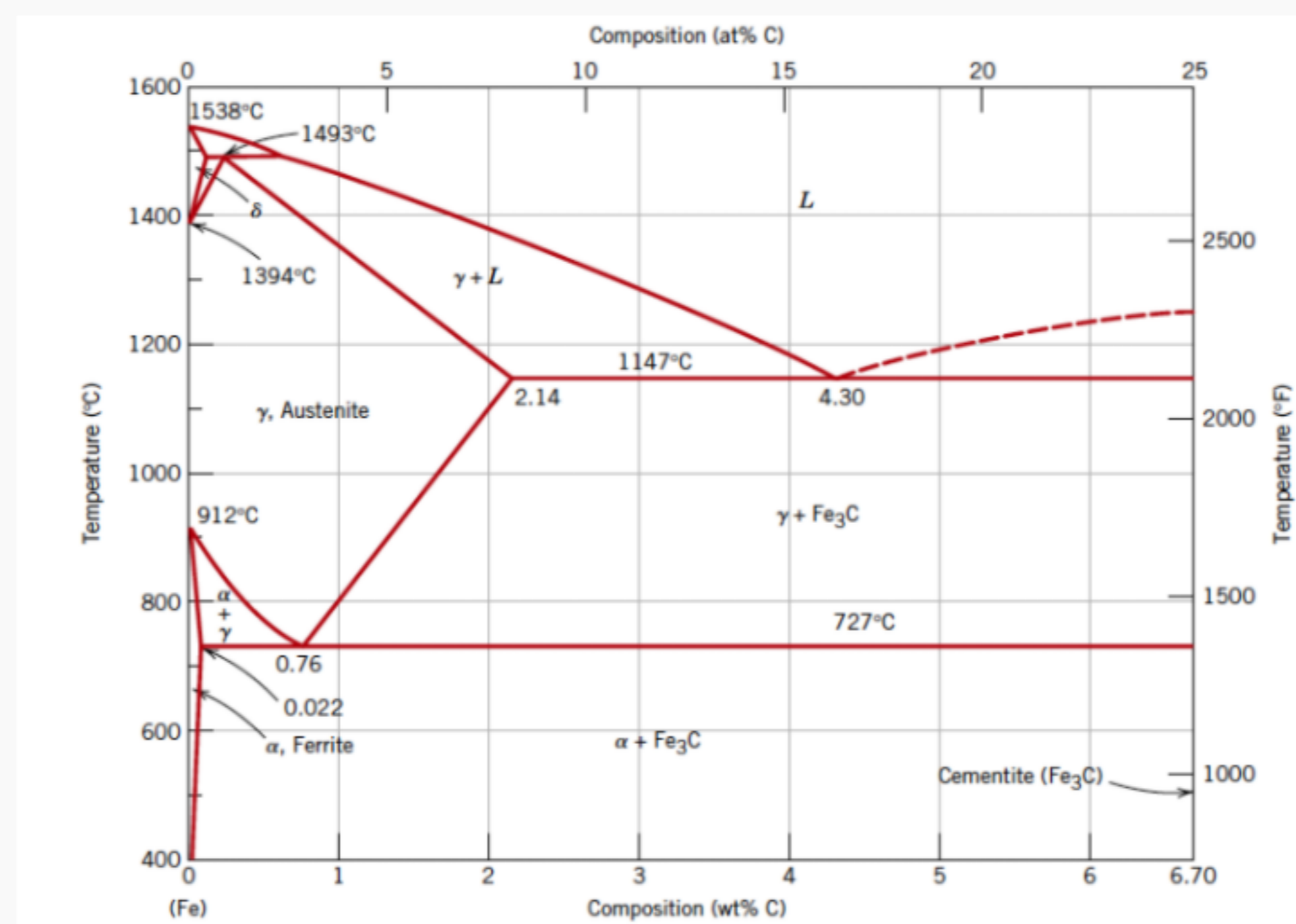


Figure 1: Fe-Fe<sub>3</sub>C Phase Diagram.

4) The eutectic mixture of austenite and cementite is called 1 point

- martensite  
 ledeburite  
 bainite  
 pearlite

No, the answer is incorrect.  
Score: 0

Accepted Answers:  
ledeburite

5) Cementite is 1 point

- a eutectic mixture of iron and carbon  
 a eutectoid mixture of iron and carbon  
 an intermediate compound of iron and carbon  
 eutectoid mixture of austenite and ferrite

No, the answer is incorrect.  
Score: 0

Accepted Answers:  
an intermediate compound of iron and carbon

6) The maximum solubility of carbon in iron in solid state is 1 point

- 4.13%  
 0.78%  
 2.14%  
 0.022%

No, the answer is incorrect.  
Score: 0

Accepted Answers:  
2.14%

7) The mass fraction of cementite in pearlite at eutectoid composition (rounded to two decimal places) is \_\_\_\_\_

No, the answer is incorrect.  
Score: 0

Accepted Answers:  
(Type: Range) 0.10,0.12

2 points

8) The carbon concentration (rounded off to two decimal places) of an iron-carbon alloy which has the total ferrite fraction as 0.80 is \_\_\_\_\_ wt% C.

No, the answer is incorrect.  
Score: 0

Accepted Answers:  
(Type: Range) 1.34,1.37

2 points

9) In a 2.5 kg of austenite containing 0.65 wt% C cooled below 727°C, the mass of pearlite (rounded off to two decimal places) is \_\_\_\_\_ kg.

No, the answer is incorrect.  
Score: 0

Accepted Answers:  
(Type: Range) 2.10,2.15

2 points

10) The SAE designation of a plain carbon steel given as 1080 can be classified as 1 point

- hypereutectoid steel  
 hypoeutectoid steel  
 hypereutectic steel  
 eutectoid steel

No, the answer is incorrect.  
Score: 0

Accepted Answers:  
hypereutectoid steel

11) The mass fraction of a eutectoid cementite in an iron-carbon alloy is 0.104. The possible composition(s) of the steel is 2 points

- 1.11 wt% C  
 0.72 wt% C  
 The data is insufficient to find the composition  
 0.2% C

No, the answer is incorrect.  
Score: 0

Accepted Answers:  
1.11 wt% C  
0.72 wt% C