



## Unit 4 - Week 1 :

# Assignment 1

The due date for submitting this assignment has passed.

**Due on 2019-08-14, 23:59 IST.**

As per our records you have not submitted this assignment.

1) The temperature of water is increased from  $-10^{\circ}\text{C}$  to  $110^{\circ}\text{C}$  keeping the pressure constant at  $0.5\text{ kPa}$ . Which among the following statements regarding the phase of water in this process is TRUE? 1 point

- (a) Ice sublimates directly from the solid to the vapour phase.  
 (b) Ice melts first into a liquid and then subsequently evaporates.  
 (c) Ice melts into a liquid and remains as a liquid till the temperature reaches  $110^{\circ}\text{C}$ .  
 (d) None of the above

- a  
 b  
 c  
 d

No, the answer is incorrect.

Score: 0

Accepted Answers:

a

2) The temperature of water is increased from  $-10^{\circ}\text{C}$  to  $50^{\circ}\text{C}$  keeping the pressure constant at  $10\text{ kPa}$ . Which among the following statements regarding the phase of water in this process is TRUE? 1 point

- (a) Ice sublimates directly from the solid to the vapour phase.  
 (b) Ice melts into a liquid and remains as a liquid till the temperature reaches  $50^{\circ}\text{C}$ .  
 (c) Ice melts first into a liquid and then subsequently evaporates.  
 (d) None of the above

- a  
 b  
 c  
 d

No, the answer is incorrect.

Score: 0

Accepted Answers:

c

3) Determine the phase of water for each of the following states. 1 point

- (1)  $100^{\circ}\text{C}$ ,  $90\text{ kPa}$   
 (2)  $100^{\circ}\text{C}$ ,  $110\text{ kPa}$   
 (3)  $100^{\circ}\text{C}$ ,  $1\text{ m}^3/\text{kg}$

- (a) (1) Superheated vapour, (2) Subcooled liquid, (3) Saturated liquid-vapour mixture  
 (b) (1) Subcooled liquid, (2) Superheated vapour, (3) Superheated vapour  
 (c) (1) Subcooled liquid, (2) Superheated vapour, (3) Saturated liquid-vapour mixture  
 (d) (1) Superheated vapour, (2) Superheated vapour (3) Subcooled liquid

- a  
 b  
 c  
 d

No, the answer is incorrect.

Score: 0

Accepted Answers:

a

4) A pressure cooker has the lid screwed on tight. A small opening with  $A = 6\text{ mm}^2$  is covered with a petcock that can be lifted to let steam escape. How much mass should the petcock have to allow boiling at  $130^{\circ}\text{C}$  with an outside atmosphere at  $100\text{ kPa}$ ? Assume  $g = 10\text{ m/s}^2$ . 1 point

- (a) 60 grams  
 (b) 102 grams  
 (c) 162 grams  
 (d) 222 grams

- a  
 b  
 c  
 d

No, the answer is incorrect.

Score: 0

Accepted Answers:

b

5) **Common data for Question 5 to 7:** Determine the missing properties for water. 1 point

$T\text{ [}^{\circ}\text{C]}$	$P\text{ [kPa]}$	$v\text{ [m}^3/\text{kg]}$	$x\text{ (if applicable)}$	Phase description
50	$P_1$	7.72	$x_1$	Phase 1

The value of  $P_1$  (in kPa) is

- (a) 7.384  
 (b) 12.350  
 (c) 19.941  
 (d) 101.3

- a  
 b  
 c  
 d

No, the answer is incorrect.

Score: 0

Accepted Answers:

b

6) The value of  $x_1$  is 1 point

- (a) 0.24  
 (b) 0.34  
 (c) 0.44  
 (d) 0.64

- a  
 b  
 c  
 d

No, the answer is incorrect.

Score: 0

Accepted Answers:

d

7) Phase 1 is 1 point

- (a) Compressed liquid  
 (b) Superheated vapour  
 (c) Saturated liquid-vapour mixture  
 (d) Cannot be determined

- a  
 b  
 c  
 d

No, the answer is incorrect.

Score: 0

Accepted Answers:

c

8) **Common data for Question 8 and 9:** Determine the missing properties for water. 1 point

$T\text{ [}^{\circ}\text{C]}$	$P\text{ [kPa]}$	$v\text{ [m}^3/\text{kg]}$	Phase description
250	400	$v_2$	Phase 2

The value of  $v_2$  (in  $\text{m}^3/\text{kg}$ ) is

- (a) 0.595  
 (b) 0.773  
 (c) 1.01  
 (d) 1.24

- a  
 b  
 c  
 d

No, the answer is incorrect.

Score: 0

Accepted Answers:

a

9) Phase 2 is 1 point

- (a) Compressed liquid  
 (b) Superheated vapour  
 (c) Saturated liquid-vapour mixture  
 (d) Cannot be determined

- a  
 b  
 c  
 d

No, the answer is incorrect.

Score: 0

Accepted Answers:

b

10) **Common data for Question 10 and 11:** Determine the missing properties for water. 1 point

$T\text{ [}^{\circ}\text{C]}$	$P\text{ [kPa]}$	$v\text{ [m}^3/\text{kg]}$	Phase description
140	500	$v_3$	Phase 3

The value of  $v_3$  (in  $\text{m}^3/\text{kg}$ ) is

- (a) 0.509  
 (b) 0.306  
 (c) 0.217  
 (d) 0.001

- a  
 b  
 c  
 d

No, the answer is incorrect.

Score: 0

Accepted Answers:

d

11) Phase 3 is 1 point

- (a) Superheated vapour  
 (b) Saturated liquid-vapour mixture  
 (c) Compressed liquid  
 (d) Cannot be determined

- a  
 b  
 c  
 d

No, the answer is incorrect.

Score: 0

Accepted Answers:

c

12) **Common data for Question 12 to 13:** Determine the missing properties for water. 1 point

$T\text{ [}^{\circ}\text{C]}$	$P\text{ [kPa]}$	$v\text{ [m}^3/\text{kg]}$	$x$
$T_4$	1000	$v_4$	0.75

The value of  $T_4$  (in  $^{\circ}\text{C}$ ) is

- (a) 151.86  
 (b) 179.91  
 (c) 198.32  
 (d) 212.42

- a  
 b  
 c  
 d

No, the answer is incorrect.

Score: 0

Accepted Answers:

b

13) The value of  $v_4$  (in  $\text{m}^3/\text{kg}$ ) is 1 point

- (a) 0.0011  
 (b) 0.0978  
 (c) 0.1461  
 (d) 0.1933

- a  
 b  
 c  
 d

No, the answer is incorrect.

Score: 0

Accepted Answers:

c

Course outline

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How to access the portal

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Data Attachment

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Week 0 Assignment 0

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Week 1 :

- Lecture 01 : Introductory Concepts
- Lecture 02 : Properties of Pure Substances
- Lecture 03 : Properties of Pure Substances (contd.)
- Lecture 04 : Introduction to Property Tables
- Lecture 05 : Properties of Pure Substances: Example problems (contd.)

Quiz : Assignment 1

Feedback for Week 1

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Week 2 :

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Week 3 :

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Week 4 :

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Week 5 :

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Week 6 :

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Week 7 :

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Week 8 :

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Week 9 :

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Week 10 :

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Week 11 :

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Week 12 :

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Assignment Solution

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Text Transcripts