Assignment 7

Due on: 2020-03-16, 23:59:57

1. What will be the activation energy for a reaction? For the reaction whose rate is constant to 30°C with a temperature in the vicinity of 30°C?

2. If a reaction is irreversible, is the activation energy for the forward reaction equal to the activation energy for the reverse reaction?

3. If a reaction is reversible, does the activation energy for the forward reaction equal the activation energy for the reverse reaction?

4. If a reaction is irreversible, does the reaction rate increase or decrease with increasing temperature?

5. If a reaction is reversible, does the reaction rate increase or decrease with increasing temperature?

6. How does the reaction rate change with increasing temperature?

7. If a reaction is irreversible, how does the reaction rate change with increasing temperature?

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