

BIOChemical & Environmental

Reaction Engineering

②

Respiration

Nitrogen fixation.

Photosynthesis.

Nitrification.

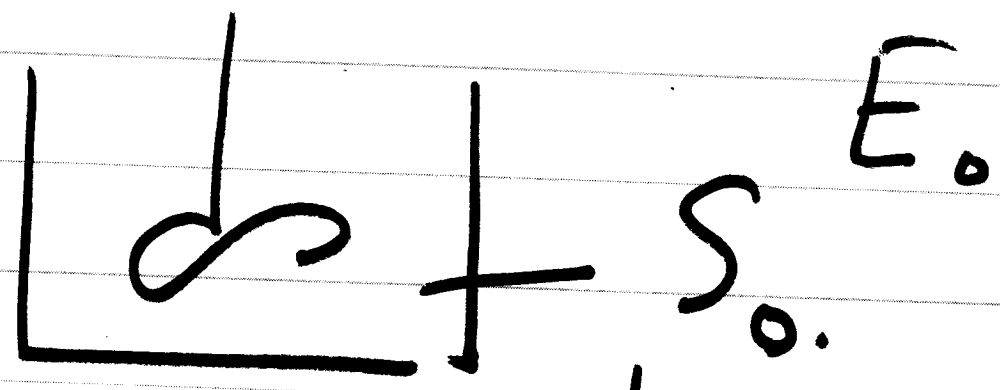
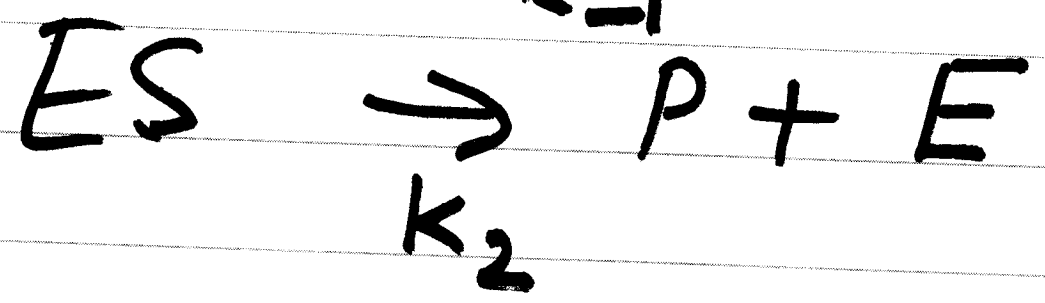
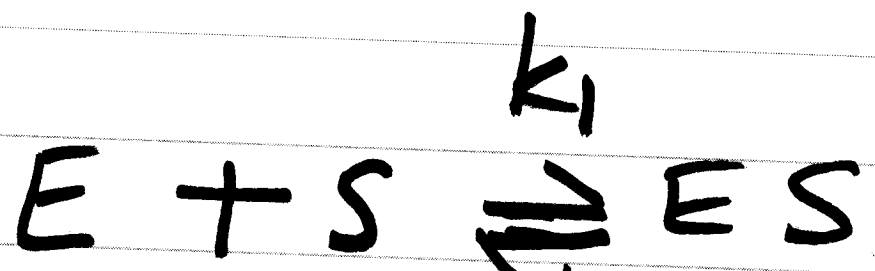
Denitrification.

# ENZ YME KINETICS

③

- 1) ENZ YMES ARE PROTEINS.
- 2) ACTIVE SITES ARISE FROM THREE DIMENSIONAL CONFIGURATION.
- 3). COFACTORS ENABLE THE ENZ YMES TO EXHIBIT THE 3D STRUCTURE

④



$$\frac{d \cdot V \cdot S}{dt} = \gamma_s \cdot V \quad \Bigg| \quad \frac{dV(es)}{dt} = \gamma_{es} \cdot V$$

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$$- \gamma_s = k_1(e)(s) - \underset{-1}{k}(es)$$

$$\rightarrow \gamma_{es} = k_1(e)(s) - \underset{-1}{k}(es) - \underset{2}{k}(es)$$

~~$$\frac{d}{dt} (v_{es}) = \gamma_{es} v$$~~

$$e_0/s_0 \ll 1$$

$$\underline{\underline{\gamma_{es} = 0}}$$

QSSA

$$e_0 = e + es \quad (6)$$

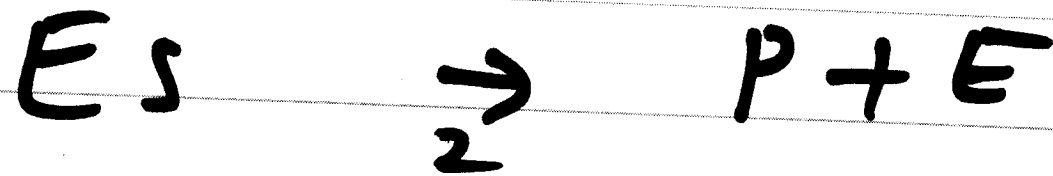
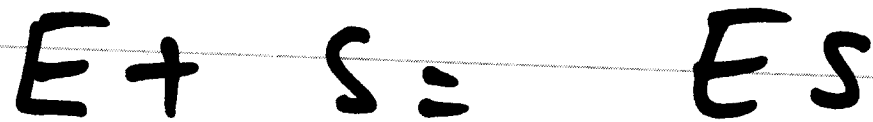
$$0 = k_1 (e)(s) - k_{-1}(es) - k_2(es)$$

$$es [k_1 + k_2] = k_1 [e - es]s$$

$$es [k_2 + k_1] = k_1 e_0 s$$

⑦

$$r_s = \frac{k_1 e_0 s}{k_{-1} + k_2 + k_1 s}$$



$$r_p = k_2 (es)$$

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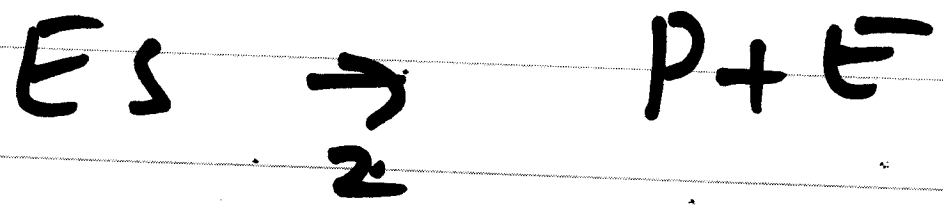
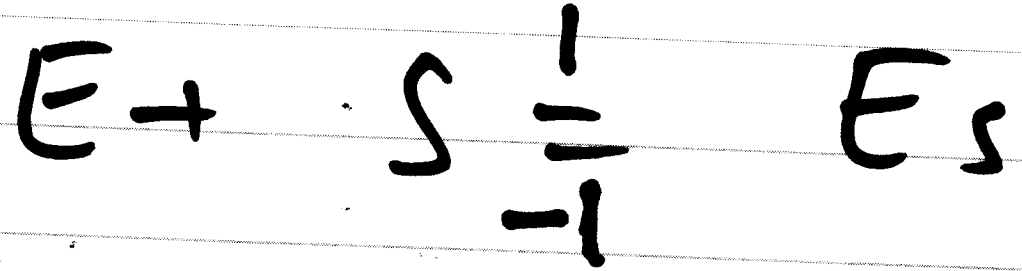
$$R_p = k_2 (e_s)$$

$$R_p = \frac{k_2 [k_1 e_0 s]}{k_{-1} + k_2 + k_1 s}$$

$$= \frac{k_2 e_0 s}{(k_{-1} + k_2)/k_1 + s}$$



⑨

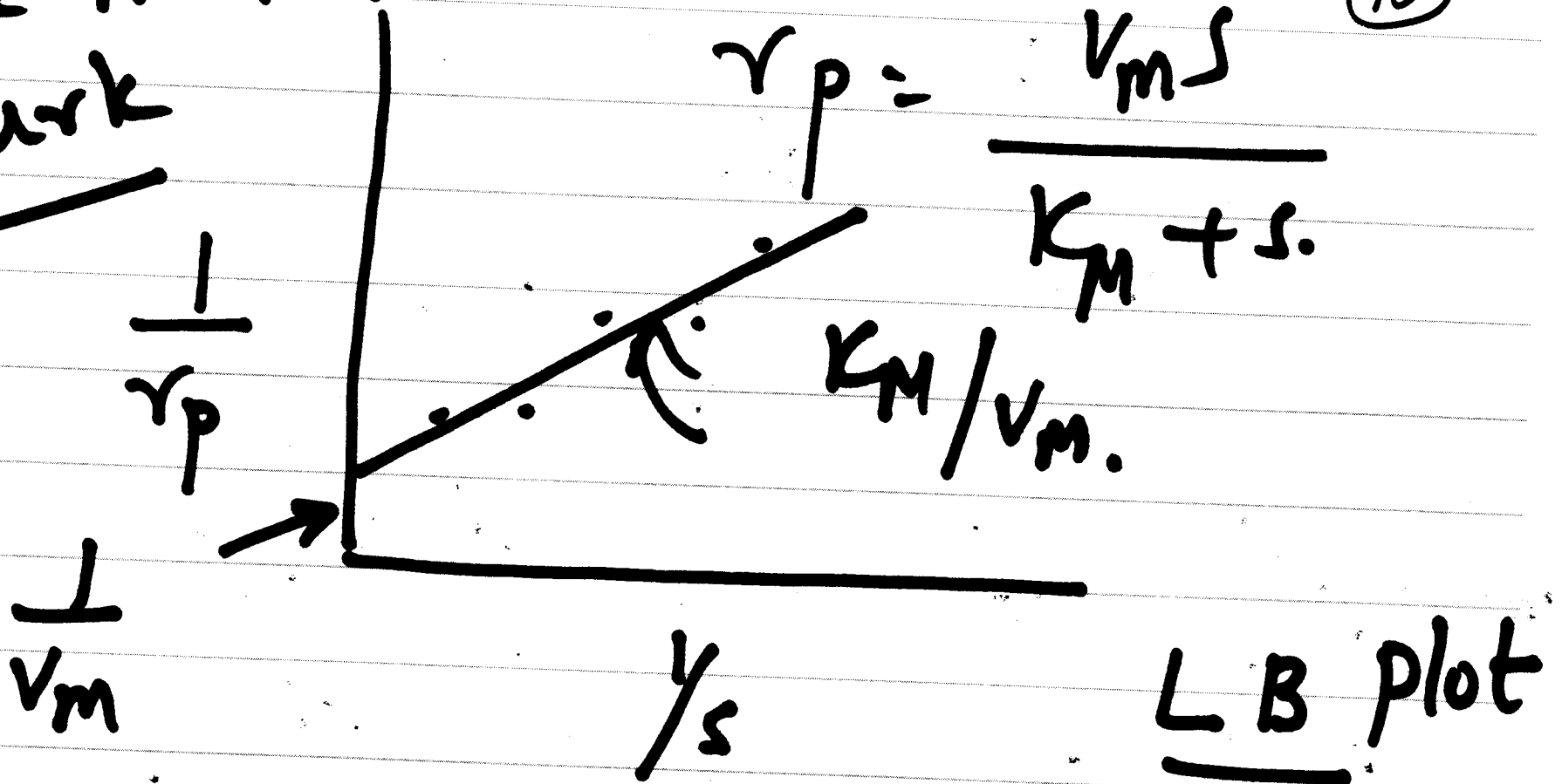


$$\left( \frac{k_1 + k_2}{k_1} \right) = K_M; \quad \underline{\underline{k_{e_0} = V_m}}$$

$$r_p = \frac{V_m S}{K_M + S}$$

# Line Weaver Burk

(10)



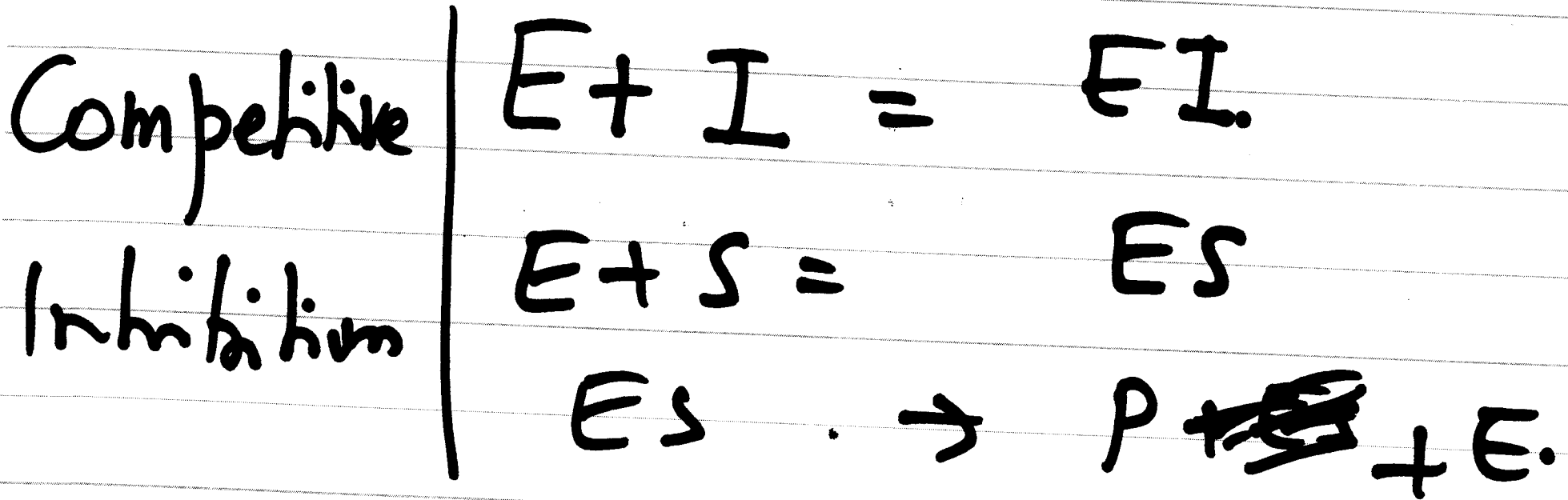
$$\frac{1}{r_p} = \frac{k_M}{v_m s} + \frac{1}{v_m}$$

②

$$\frac{1}{r_p} = \frac{K_M}{V_{mS}} + \frac{1}{V_m}$$

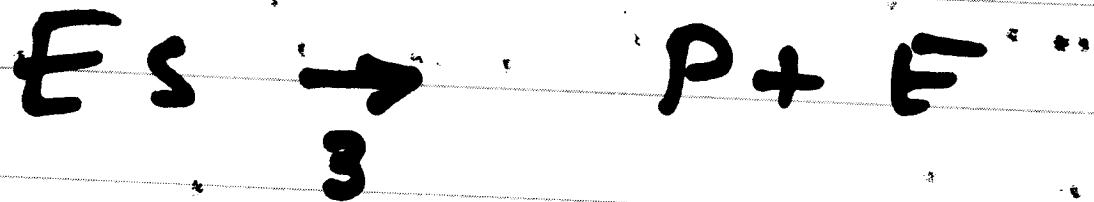
$$r_p = \bar{r}_p + \epsilon$$

$$r_p = \frac{V_m S}{K_m + S}$$



$$S + E \xrightarrow{1} Es$$

$$I + E \xrightleftharpoons[-2]{} EI$$



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$$0 = \cancel{r_{eS}} = k_1(e)(s) - k_2(es) - k_3(es)$$

$$0 = \cancel{r_{eI}} = k_2(e)(I) - \underset{-2}{k_1(eI)}$$

$$(es) \begin{bmatrix} k_1 + k_3 \\ -1 \end{bmatrix} = k_1 \begin{bmatrix} e_0 - es - eI \\ \uparrow \end{bmatrix}$$

$$-Y_s = \frac{k_1(e)(s) - k_{-1}(es)}{k_3}$$

$$k_2 [e_0 - e_s - e_1] I = k_2 (eI)_{-2}$$

~~$$(eI) = [e_0 - e_s - e_1] k_2$$~~

~~$$(eI) = (e_0 - e_s - e_1)(k_2)$$~~

~~$$k_2 [e_0 - e_s - e_1]$$~~



(17)

$$k_2 [e_0 - e_s - e_1] I$$

$$= k (e_1)$$

$$\cancel{e_1} \left[ \begin{matrix} k_1 + k_2 I \\ -2 \end{matrix} \right] = \frac{k I}{2} [e_0 - e_s]$$

$$e_1 = \frac{k}{2}$$

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$$g_{e1} = v = k_2 [e_0 - e_1 - es] I$$

$$- (e_1) k_{-2}$$

$$e_1 [k_2 I + k_{-2}] = k_2 I [e_0 - es]$$

$$0 = \cancel{g_{es}} = k_1(e)(s) - \underset{-1}{k_2(es)} - \underset{-1}{k_3(es)}$$

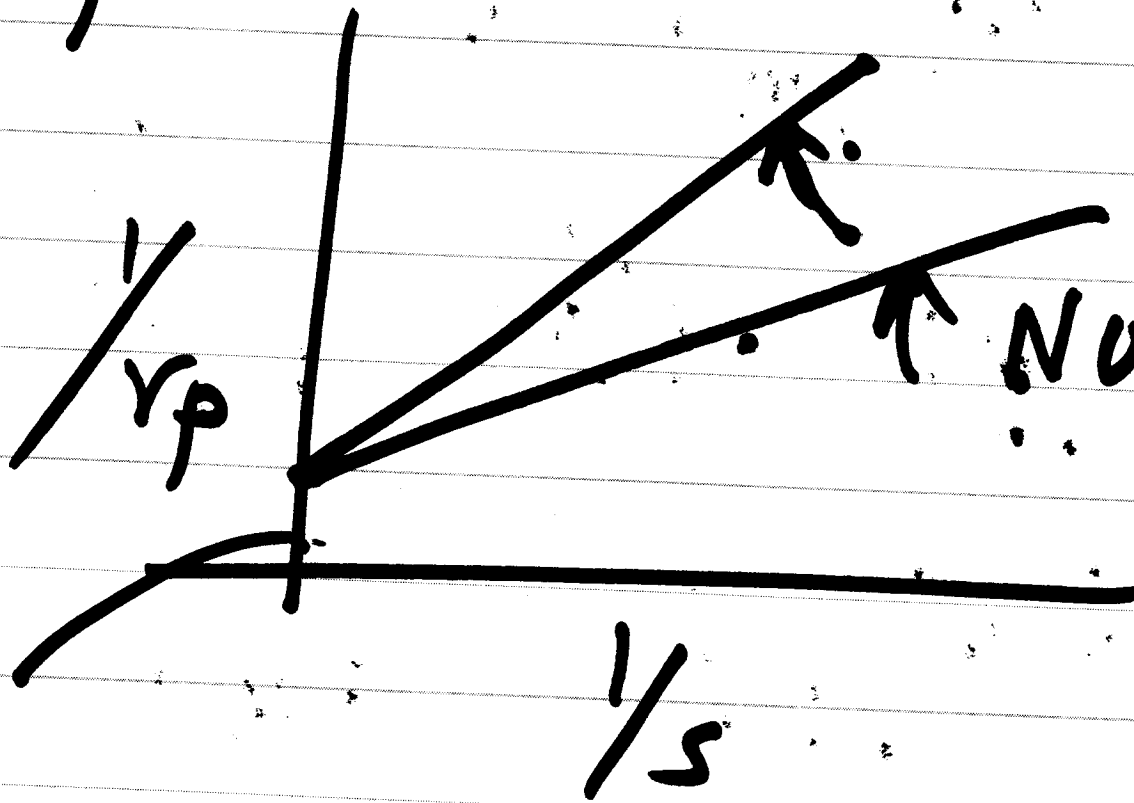
$$0 = k_1(e_0 - es - eI) - \underset{-1}{k_2(es)} - \underset{-1}{k_3(es)}$$

$$G_p = \frac{V_M s}{s + K_M \left[ 1 + \frac{I}{K_i} \right]}$$

$$K_i = \frac{k_{-2}}{k_2} \quad \begin{array}{l} E + S = ES \\ E + I = EI \end{array}$$

$$K_M = \frac{(k_{-1} + k_3)}{k_1}$$

$$\frac{1}{r_p} = \frac{1}{v_m} + \frac{k_m}{v_{ms}} \left[ 1 + \frac{I}{k_i} \right] \quad (2)$$



COMP.

NO I.

UN COM

$$E + S = ES$$

$$I + ES = IES$$

$$ESF \Rightarrow P + E$$

$$r_p = \frac{VMS}{K_m + S(1 + I/K_i)}$$

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COM.

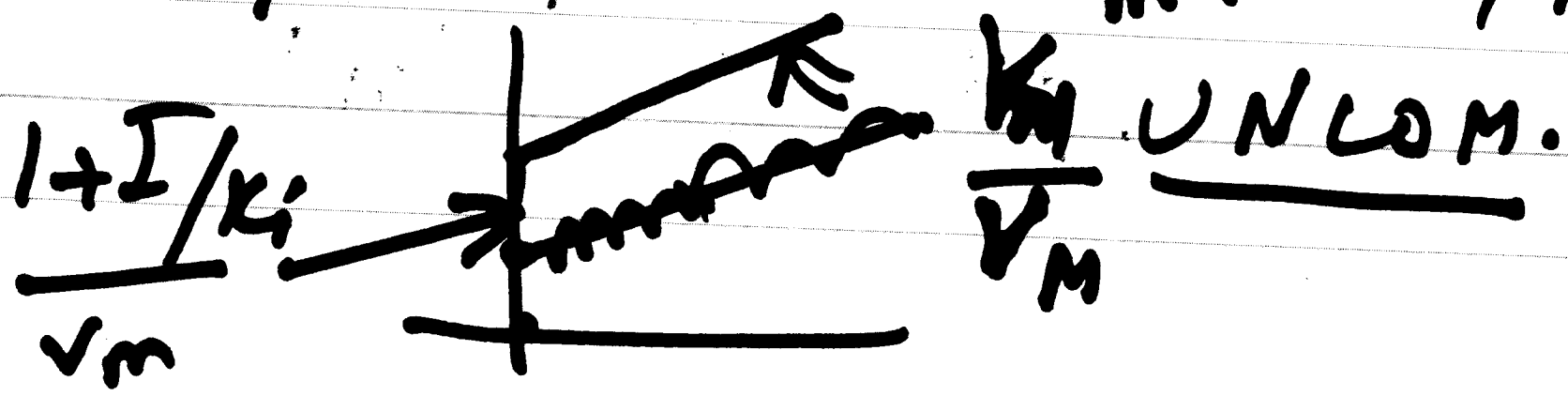
$$E + S = ES$$

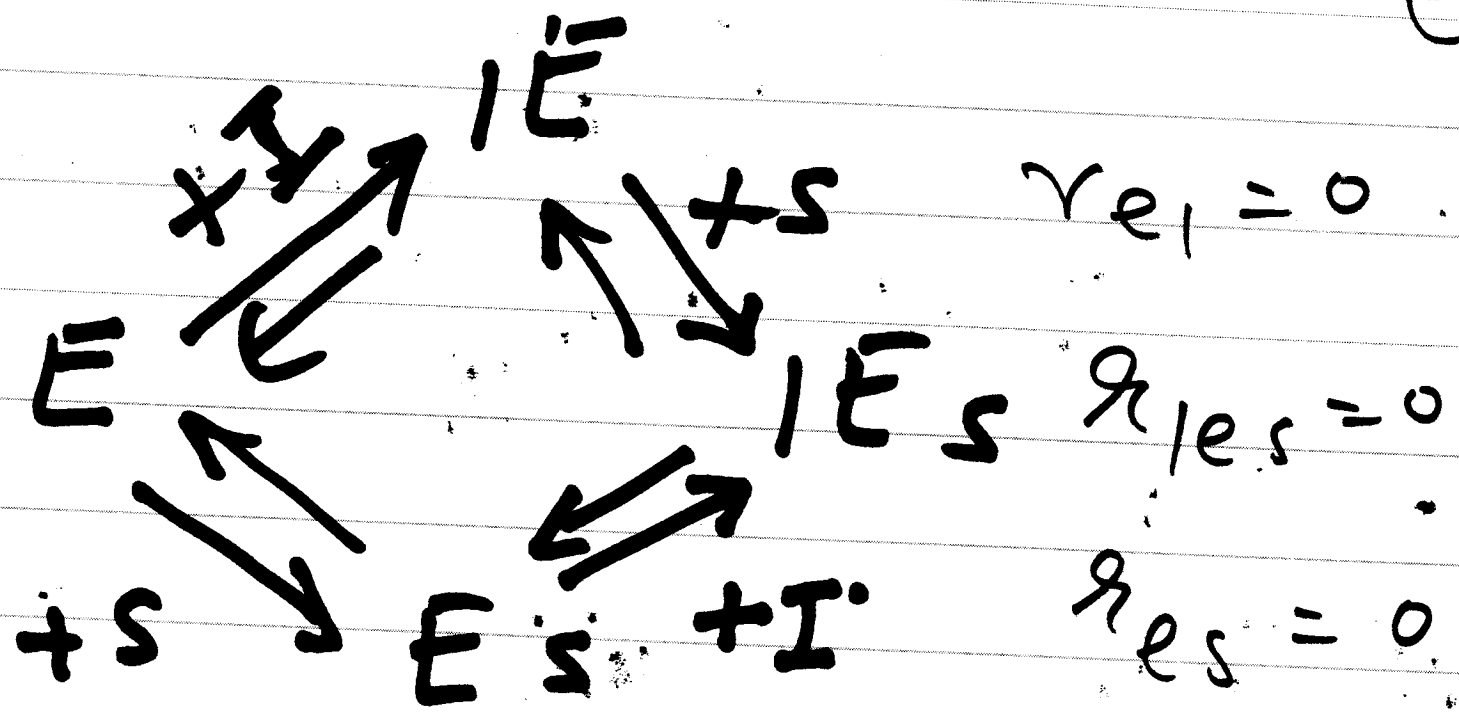
$$E + I = EI$$

$$ES \Rightarrow P + E$$

$$r_p = \frac{V_m s}{K_M + s \left( 1 + \frac{I}{K_i} \right)}$$

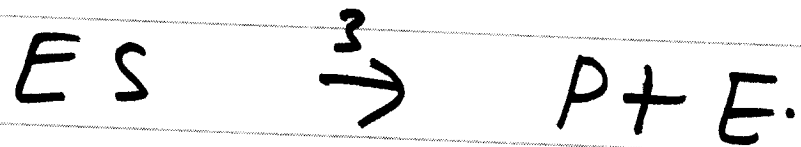
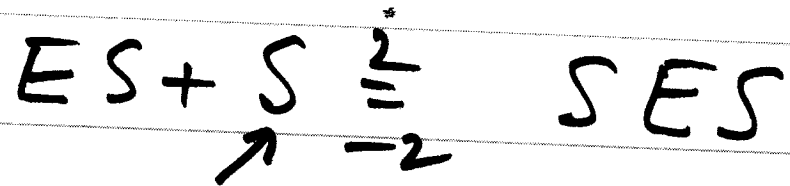
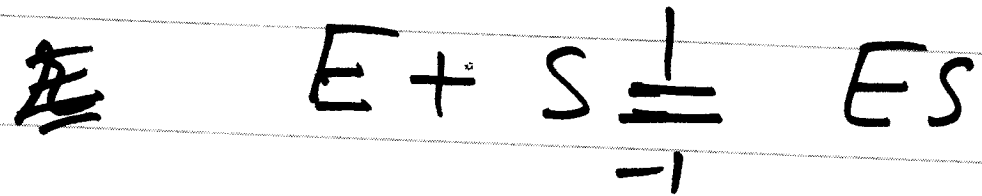
$$\frac{1}{r_p} = \frac{K_M}{V_m s} + \frac{1 + I/K_i}{V_m (1 + I/K_i)}$$





$$\gamma_p = \frac{V_m s}{(s + K_m)(1 + I/K_i)}$$





$e_0 = e + es + ses$

$0 = \frac{d}{dt} es = k_1(e)(s) - k_{-1}(es) + k_2(ses)$

$0 = \frac{d}{dt} ses = -k_{-2}(s)(es) - k_3(es) + k_2(es)(s) - k_{-2}(ses)$

$$r_p = \frac{k_3 (e_0) k_1 s}{k_1 s + \left( \frac{k_{-1} + k_3}{k_1} \right) + \frac{k_1 k_2 s^2}{k_2}}$$

$$k_1 s + \left( \frac{k_{-1} + k_3}{k_1} \right) + \frac{k_1 k_2 s^2}{k_2}$$

~~$r_p = \frac{k_3 e_0 s}{k_1 s + \left( \frac{k_{-1} + k_3}{k_1} \right) + \frac{k_1 k_2 s^2}{k_2}}$~~

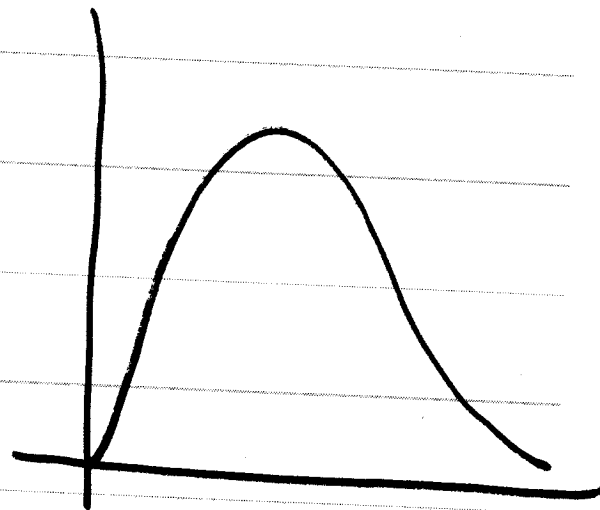
~~$\left[ s + \left( \frac{k_{-1} + k_3}{k_1} \right) + \frac{k_2 s^2}{k_2} \right]$~~

$$r_p = \frac{V_M s}{1 + \frac{k_2}{s^2} + \frac{s k_1}{k_1}}$$

$$\left( \frac{dr_p}{ds} \right) = 0$$

$$k_1 = \left[ \frac{k_{-2}}{k_2} \right]$$

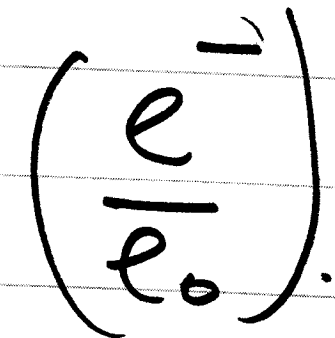
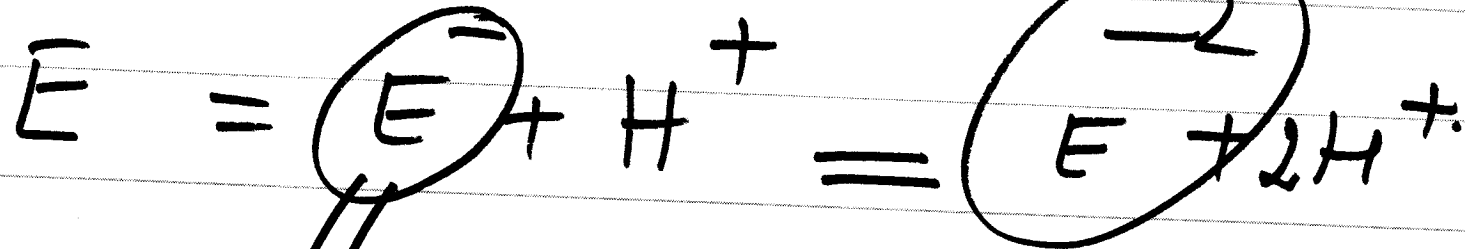
$$k_2 = \left[ \frac{k_{-1} + k_3}{k_1} \right]$$



$$\frac{dr_p}{ds} = 0.$$

$$s_{opt} = \sqrt{K_1 K_2}$$

# pH ~~Re~~ EFFECTS.



$$K_1 = \frac{e^- H^+}{e}$$

$$K_2 = \frac{e^{-2} (H^+)^2}{(e^-) (H^+)^2}$$

$$e_0 = e + e^- + e^{-2}$$

$$e_0 = \frac{e^- H^+}{K_1} + e^- + K_2 (e^-) / H^+$$

$$e^- \left[ \frac{H^+}{K_1} + 1 + \frac{K_2}{H^+} \right] = e_0$$

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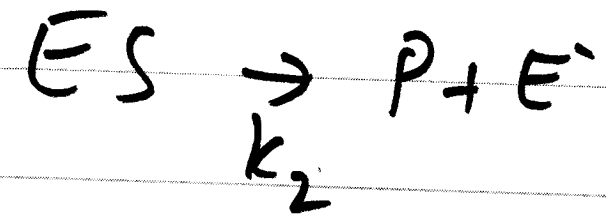
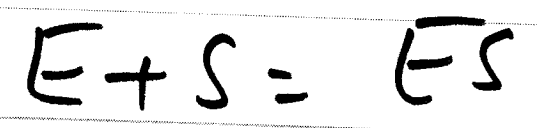
$$\left( \frac{e^-}{e_0} \right) = \frac{1}{\left[ 1 + \frac{H^+}{K_1} + \frac{K_2}{H^+} \right]}$$

~~H<sup>+</sup>~~

$$\left( \frac{e^-}{e_0} \right)$$

$$H^+ = \sqrt{K_1 K_2}$$

$$r_p = \frac{V_M S}{K_M + S}$$



(H<sup>+</sup>)

$$V_M = \left( \frac{k_2 P_0}{k_1} \right) (y)$$

$$y = \frac{e}{e_0}$$



# MICROBIAL REACTION.

$$r_x = \mu x$$

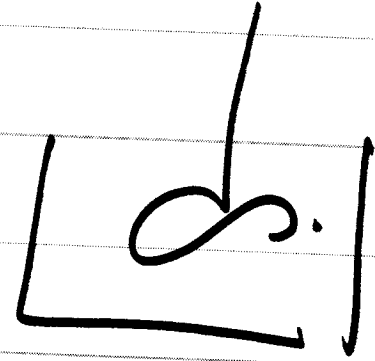
$\mu =$  sp. Growth Rate

$$\mu = \frac{\mu_m S}{K_s + S}$$

$K_s$ : MONOD CONSTANT.

$\mu_m$ : MAX. GROWTH RATE.

$$v \frac{dx}{dt} = q_x \cdot v$$



$$v \frac{ds}{dt} = q_s \cdot v$$

$$\left( \frac{dx}{ds} \right) = \frac{q_x \leftarrow}{q_s \leftarrow} = -(\gamma)$$

$$x - x_i = \gamma (s_i - s)$$

$$\frac{dx}{ds} = \frac{r_x}{r_s} = -\gamma$$

$$(x - x_i) = \underbrace{(\gamma)}_{\text{Yield Coeff}} [s_i - s]$$

Yield Coeff

$$\frac{dx}{dt} = \downarrow \mu \uparrow \frac{dx}{dt}$$

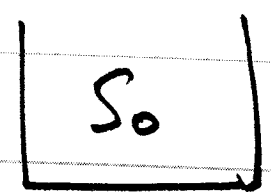
$$\left( \frac{dx}{dt} \right) = \left[ \begin{matrix} \mu \\ \uparrow \end{matrix} x \right]$$

$$\frac{dx}{dt} = \left[ \begin{matrix} \mu_m \downarrow s \\ K_s + s \end{matrix} \right] \left[ \begin{matrix} x \\ \uparrow \end{matrix} \right]$$

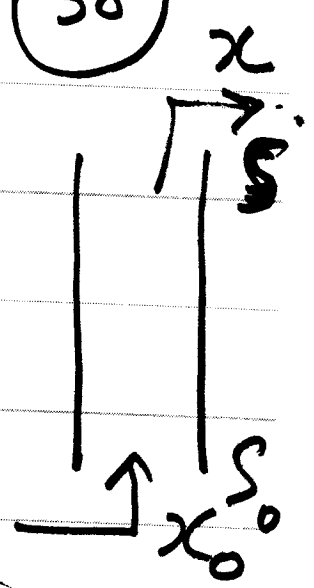
(37)

$$\frac{d}{dt} \begin{bmatrix} y(s_i - s) \\ + x_i \end{bmatrix} = \frac{\mu_m s.}{K_s + s} \begin{bmatrix} x_i \\ + y(s_i - s) \end{bmatrix}$$

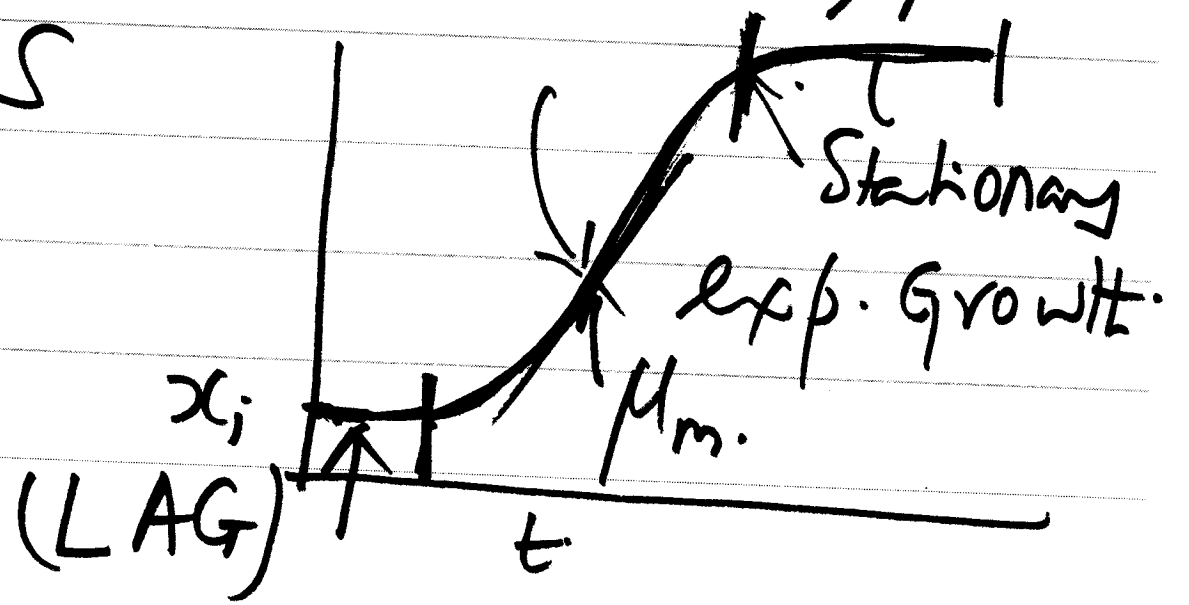
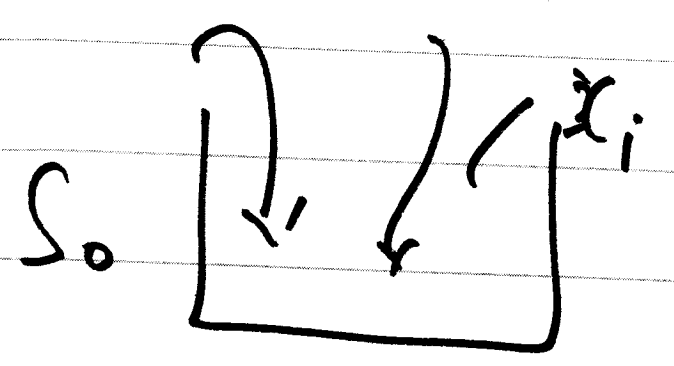
$$- \frac{y ds.}{dt} = \frac{\mu_m s.}{(K_s + s)} \begin{bmatrix} x_i + y [s_i - s] \end{bmatrix}$$

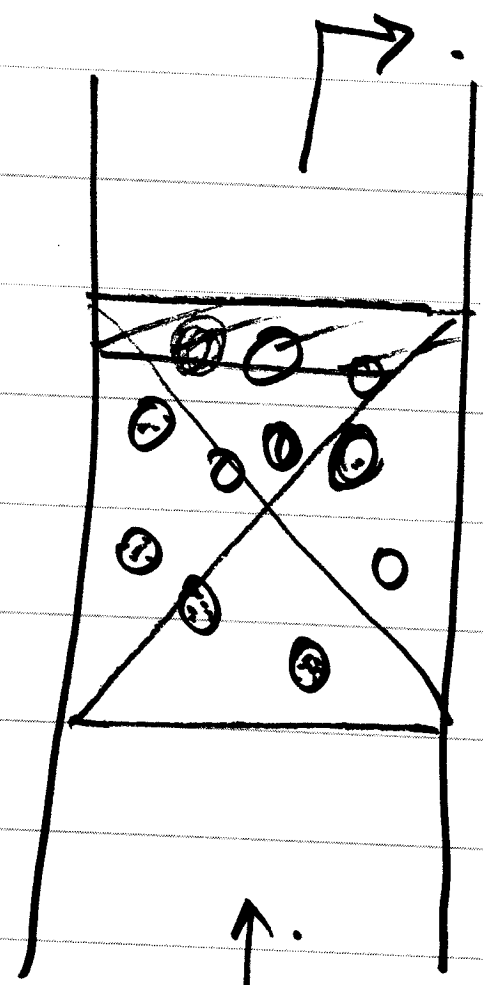
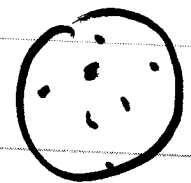


$$(x_i + \gamma K_s + \gamma S_i) \ln \left[ \frac{x_i + \gamma S_i - \gamma S}{x_i} \right]$$



$$= \gamma K_s \ln \frac{S_i}{S} = (x_i + \gamma S_i) / \mu_m t.$$





IM

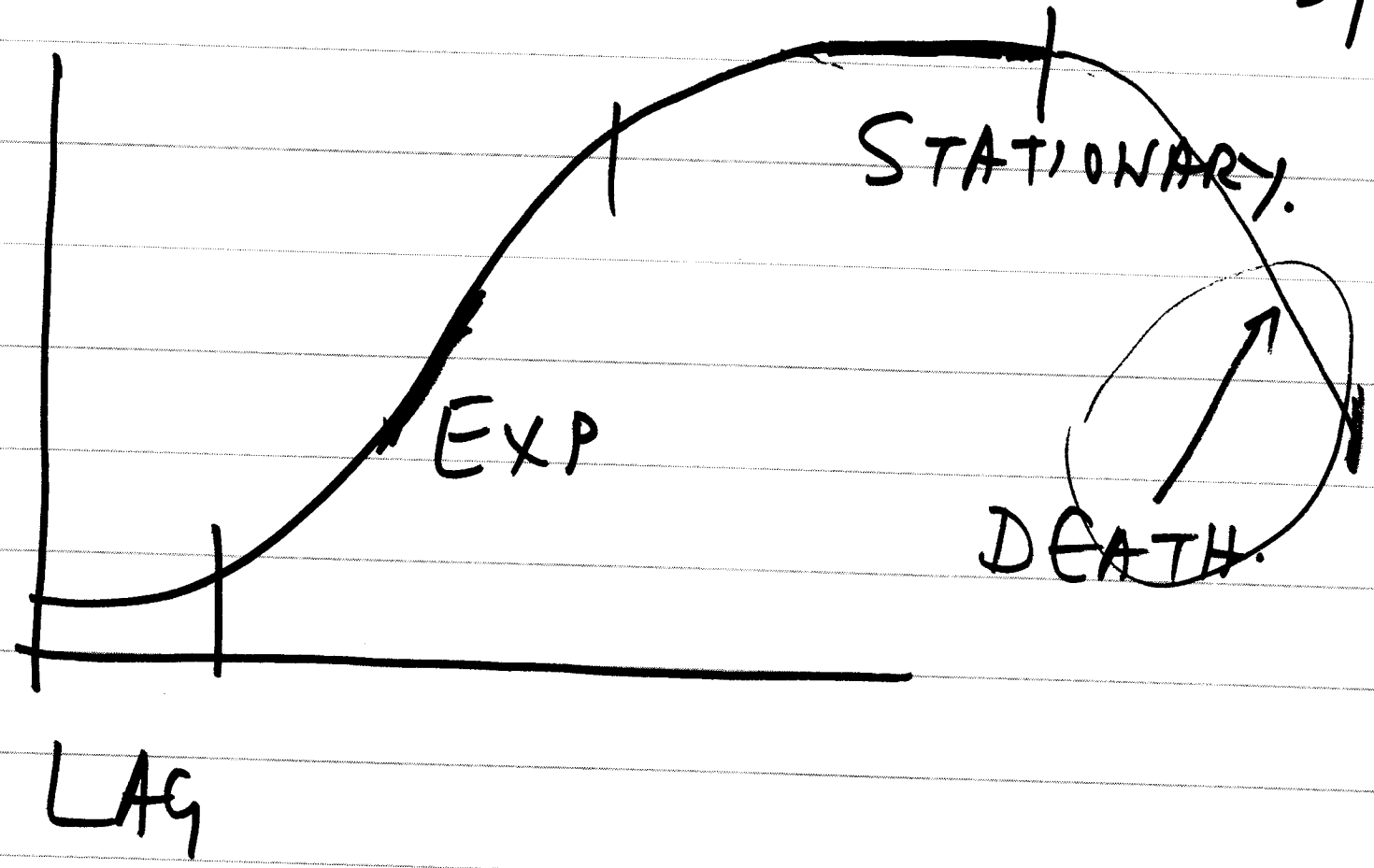
○ IMM. MICRO  
PELLETS.

$$(x_i) + x_0 + \gamma(s_0 - s)$$

$x_i$



$s_0$   
 $(x_0)$





ICR

$$(\downarrow x_0 + \downarrow x_i + \gamma K_s + \gamma S) \ln \left[ \frac{x_0 + x_i + \gamma S_0 - \gamma S}{x_0 + x_i} \right]$$

$$+ \gamma K_s \ln \frac{S_0}{S}$$

$$= (x_0 + x_i + \gamma S_0) \underbrace{(1 - \epsilon_s - \epsilon_g)}_{\downarrow \downarrow} \left( \frac{1}{\mu_m} \right)^{\downarrow} \tau. \quad \downarrow \text{V/F.}$$

ALCOHOL

## PLANT G FLOW

$$\frac{dx}{dv} = r_x = \mu x$$

$$\frac{dx}{(1-\epsilon_g - \epsilon_e)dv} = \frac{\mu_m \left[ x_i + x_0 + \gamma (S_0 - s) \right]}{K_s + s}$$

$$V_e = V \left[ 1 - \epsilon_g - \epsilon_e \right]$$