

Prof. Shankar
Lec. ~~3012~~ 4/12/2012

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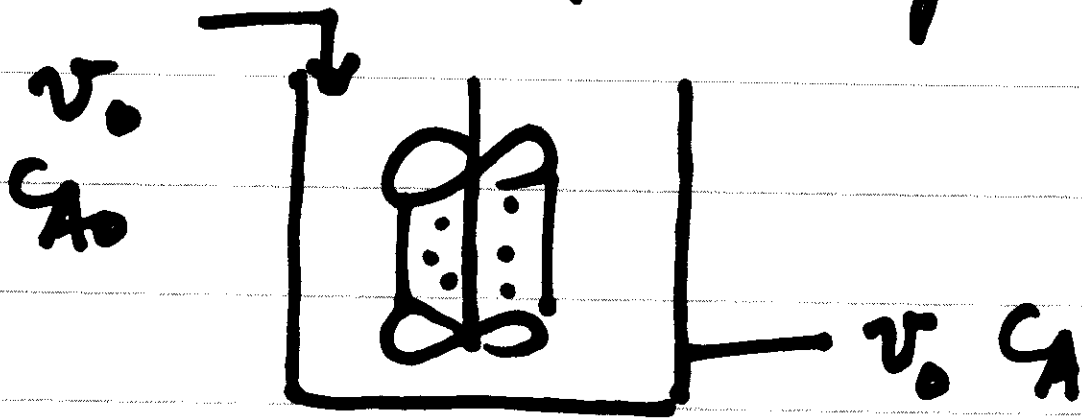
Q 12

Advanced Reaction Engineering

Time Dependent operations

4/12

Deactivating Catalysts



$$v_0 C_{A0} - v_0 C_A + r_A V = V \frac{dC_A}{dt}$$

$$\psi = C_A / C_{A0}; \quad \theta = t / \tau_d$$

τ_d : Characteristic Time for deactivation

$$1 - \psi + \frac{g_A V}{v_0 C_{A0}} = \frac{v}{\tau_d} \frac{d\psi}{d\theta} v_0$$

$$1 - \psi + \gamma_A \frac{\tau_r}{C_{A0}} = \frac{\tau_r}{\tau_d} \frac{d\psi}{d\theta}$$

$$\tau_r \ll \tau_d$$

4

A → Products

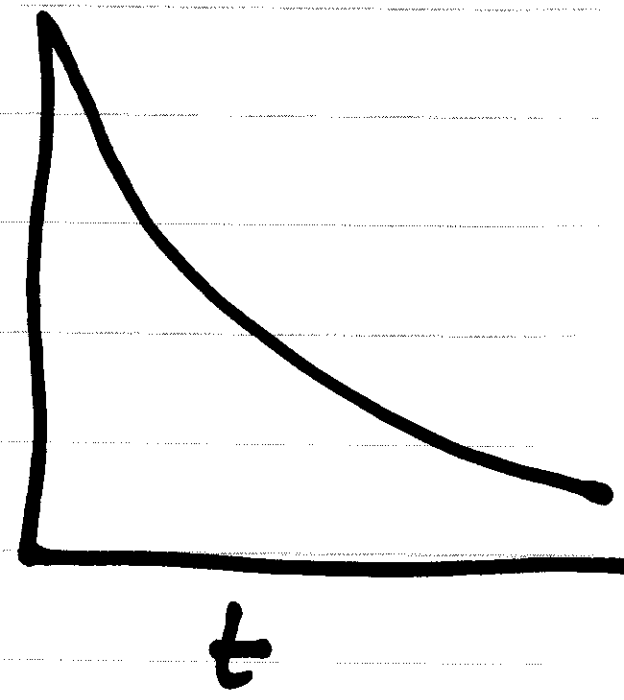
$$1 - \psi + r_A \frac{\tau_r}{\tau_0} = 0 \quad \text{QSSA}$$

$$(-) r_A(t) = [1 - \psi(t)] \frac{C_{A0}}{\tau_r}$$

mini
video

$\psi(t); r_A(t)$

$-r_A$



5

$$a = \frac{r_A(t)}{r_A(0)}$$

↑
r_A(0)

Standard state

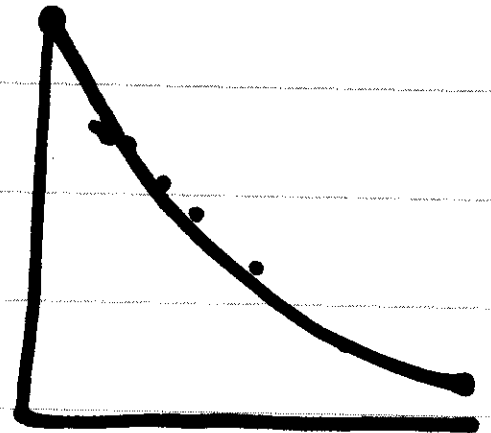
$$r_A(t) = (a) r_A(0)$$

miss
in video

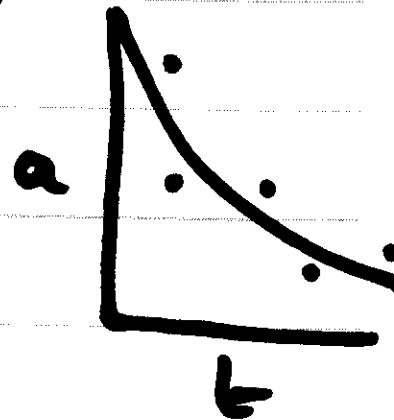
$$\Rightarrow (-) k(\tau) f(c) (a)$$

$$r_d = k_d a^m c_A^n c_B^p$$

-r_A



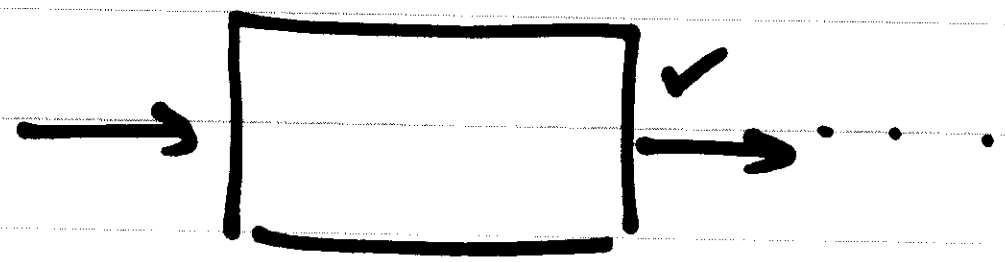
t



6

$$-r_A = \underbrace{k(T) \cdot (a)}_k \cdot \underbrace{f(c)}_f$$
$$r_d = k_d a^m \cdot C_A^n \dots$$

$$-\frac{da}{dt} = r_d$$



$$k(T) a \{ \dots \} = \text{constant}$$

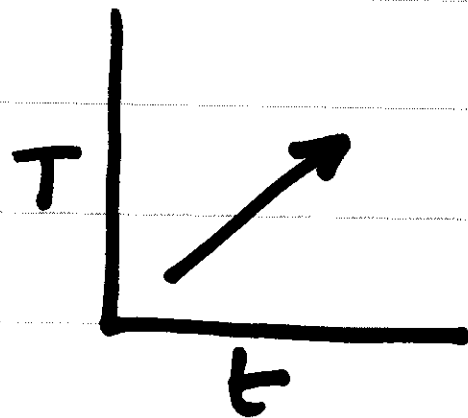
7

$$k(T) \cdot a(T, t, G, \dots) = \text{Constant}$$

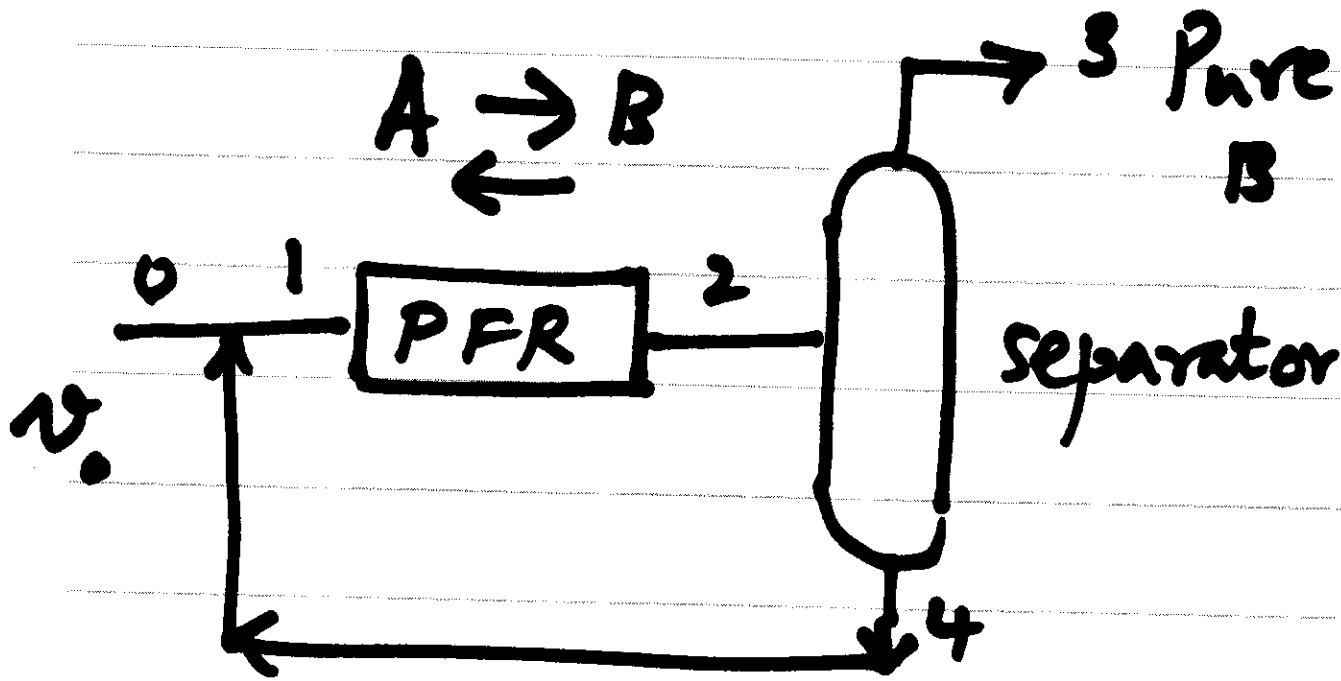
$$a = A e^{-k_d t}$$

$$k_d = k_{d0} \exp(-E_d/RT)$$

$$k(T) = A_0 \exp[-E/RT]$$



8

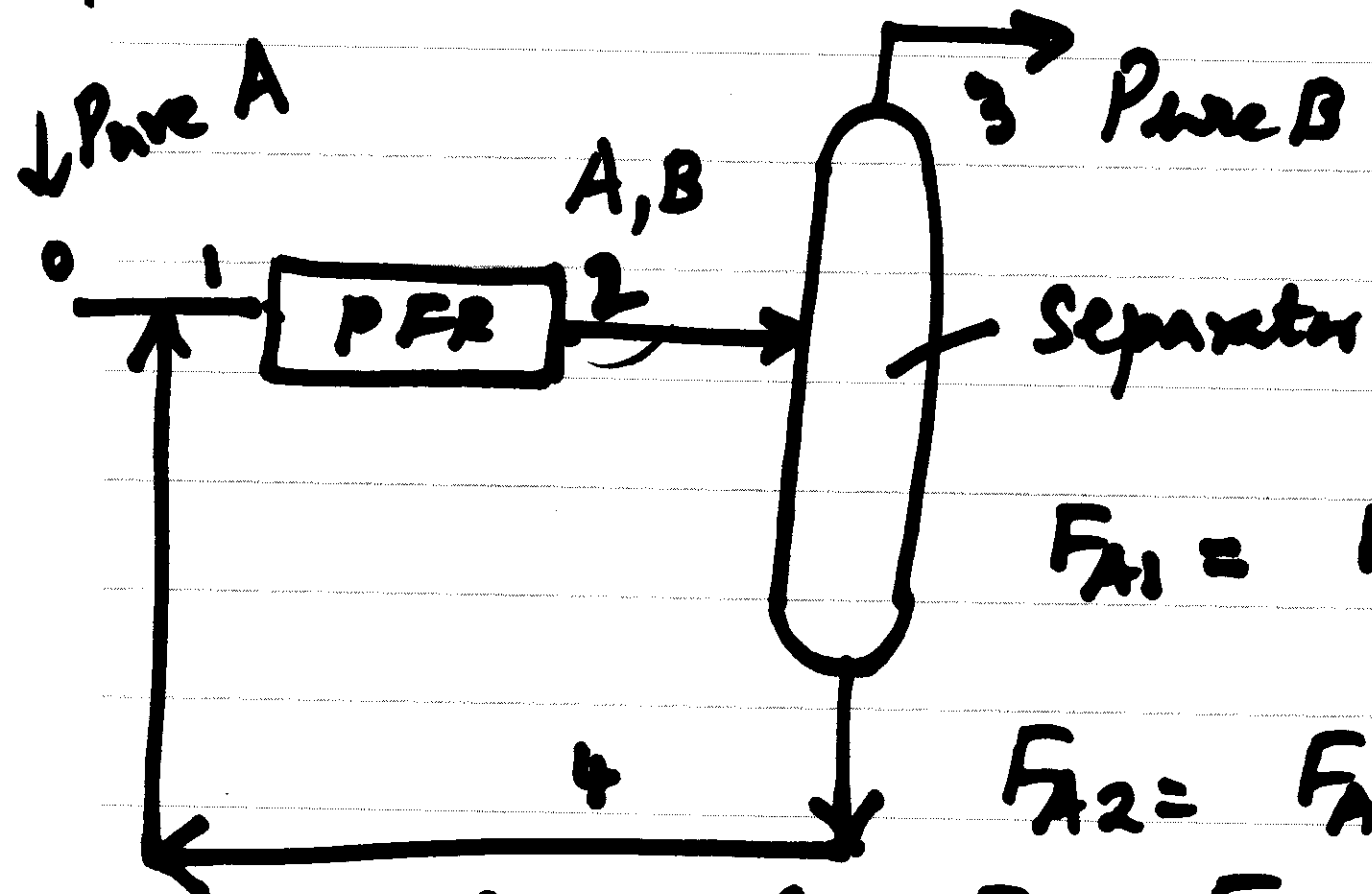


$$k = \exp\left[8.8 - \frac{5000}{T(K)}\right]$$

$$k_{12} = \exp\left[1.3 - \frac{2500}{T(K)}\right]$$

$$K = \exp\left[-19.5 + \frac{10000}{T(K)}\right].$$

Catalyst	T K.	Feed F _{A0} lit/s	A at Reactor exit %
Fresh	450)	0.39	25
Spent	477	0.38	32.2
Regenerated	450	0.28	19



$$F_{A1} = F_{A0} + F_{A4}$$

$$F_{A2} = F_{A1} (1 - y_2) = F_{A4}$$

$$F_{A4} = \text{Pure A} = F_{A0} \quad F_{A1} = F_{A0} + F_{A1} (1 - y_2) \Rightarrow \frac{F_{A0}}{y_2}$$

y_2 is conversion at position 2 with respect to F_{A1} at position 1.

"

$$C_{A4} = C_{A0} \Rightarrow C_{A1} = C_{A0}$$

$$v_4 = \frac{F_{A4}}{C_{A4}} = \frac{F_{A2}}{C_{A4}} = \frac{F_{A1}(1-y_2)}{C_{A0}}$$

$$= \frac{F_{A0}(1-y_2)}{y_2 C_{A0}} = \frac{v_0(1-y_2)}{y_2}$$

$$v_1 = v_0 + v_4 = v_0 + \frac{v_0(1-y_2)}{y_2} = \frac{v_0}{y_2}$$

$$F_{A1} = F_{A0}/y_2$$

12

Design Eqn PFR

$$\frac{dF_A}{dV} = r_A$$

$$C_{A2} = C_{A0}(1-y_2)$$

$$C_{B2} = C_{A0} y_2$$

$$C_A(\text{at any position}) = C_{A0}(1-y) = \frac{F_{A0} \cdot (1-y_2) y_2}{v_0}$$

$$C_B(\dots) = C_{A0} y = y_2 v_0 C_{A0}(1-y_2).$$

$$C_{A1} = C_{A0}.$$

$$C_{A2} = \frac{F_{A2}}{v_2} = \frac{F_{A1}(1-y_2)}{v_1}$$

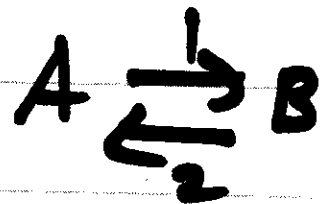
$$= \frac{F_{A1}(1-y_2)y_2}{v_0}$$

$$= \frac{F_{A0} \cdot (1-y_2) y_2}{v_0}$$

$$= y_2 v_0 C_{A0}(1-y_2).$$

$$F_A = F_{A_1} (1-y)$$

$$\frac{dF_A}{dv} = k_2 \times C_B - k_1 \times C_A$$



$$- F_{A_1} \frac{dy_E}{dv} = k_2 \times C_{A_0} y - k_1 \times C_{A_0} (1-y) \quad \alpha - \text{activity}$$

$$\frac{F_{A_0}}{y_2} \frac{dy_E}{dv} = k_{(1)} \times C_{A_0} (1-y) - k_{(2)} \times C_{A_0} y$$

$$\frac{v_0}{y_2} \frac{dy}{dv} = k_{(1)} (1-y) - k_{(2)} y$$

$$\frac{v_0}{y_2} \frac{dy}{dv} = k_{(1)} \alpha (1-y) - k_{(2)} \alpha \beta y.$$

$$(k_{(1)} \alpha v) = - \frac{v_0}{y_2 \beta} \ln(1 - \beta y_2)$$

$K_e = \text{Eq. Constant}$

$$\beta = (K_e + 1) / K_e$$

15

Fresh Catalyst

$$k_{(1)} \propto V = - \frac{U_0 \ln(1 - \beta y_2)}{y_2 \beta}$$

$$\beta = \frac{K_e + 1}{K_e} = \frac{8.5 + 1}{8.5}$$

$K_e(450K) = 8.5$ from data given

Fresh Catalyst Contd.

$$k_1(\alpha)V = \frac{-0.39}{0.75} \ln [1 - 1.117 \times 0.75]$$

$$(k_1 \alpha V)_F = 0.847 \quad (\text{fresh catalyst}).$$

2/12

Mol. Fraction at port 2 of A

$$\frac{F_{A1} (1-y_2)}{F_{A1} (1-y_2) + F_{A1} y_2} = (1-y_2) = 0.25$$

~~1 = 0.25~~ ~~0.75~~

$$1 - y_2 = 0.25$$

$$y_2 = 0.75$$

24/18

Spent: 477 K

Eq. Constant at 477K. = 6.

$$\beta = (6+1)/6 = 1.16$$

$$\begin{aligned} (k_{(1)} \propto V)_{477} &= -\frac{v_0}{\beta y_2} \ln(1 - \beta y_2) \\ &= \frac{-0.38}{1.16(0.678)} \ln(1 - 1.16 \times 0.678) \end{aligned}$$

$$(k_{(1)} \propto V)_{477} = 0.75$$

23 19

$$\frac{(k, \alpha V)_{477}}{(k, \alpha V)_{450}} = \frac{0.75}{0.847}$$

$$(k, \alpha V)_{450} = 0.88$$

$$\frac{(k, \alpha)_{477}}{(k, \alpha)_{450}} = (0.88)$$

$$(k, \alpha)_{450}$$

Activity of spent catalyst at

$$477 \text{ K} = (0.88) \frac{k_{0,}(450 \text{ K})}{k_{0,} \text{ at } 477 \text{ K.}}$$

Spent Cat.

$k_{0,}$ at 477 K.

$$\frac{(\alpha)_{477}}{(\alpha)_{450}} = \frac{0.88}{2.16} = (0.33)$$

$(\alpha)_{450}$

$$(2.25)(0.33) \approx \underline{\underline{0.71}}$$

21

Regenerated Catalyst.

$$(k_{(1)} \alpha V)_{Reg} = \frac{-v_0 \ln(1 - \beta y_2)}{y_2 \beta}$$

$$= \frac{-0.28 \ln(1 - 1.117 \times 0.81)}{(0.81)(1.117)}$$

$$(k_{(1)} \alpha V)_{Reg} = 0.727$$

Req.

$$\frac{(k_1 \alpha V)_{Req} (450)}{(k_1 \alpha V)_{Fuel} 450} = \frac{0.727}{0.847} = \underline{\underline{0.858}}$$

Reg. Cat. amt $>$ 2.25 km Spmt

$$\underline{0.727} > \underline{0.71}$$

S. critery Π is satisfied.